

Physical Science 1st Hour

September 8 – September 12

Monday, September 8

States of Matter

Standards

PSc.2.1.2 - North Carolina

Explain the phases of matter and the physical changes that matter undergoes.

Essential Questions

How do I tell the difference between the states of matter? [solid, liquid, gas and plasma]

Warm Up

Is it melting? Formative Assessment Probe

Objective

Scholars will

PSc 2.1.1 Identify the states of matter [solid, liquid, gas, plasma]

PSc 2.1.2 Understand and diagram the physical changes of matter.

Activity 1

Discuss Changes in matter with guided notes

Activity 2

Flow Map of Physical Changes of Matter

Activity 3

Understand the energy graph needed to have change in state of matter.

Assessment

Formative Probe = Is it Melting?

Study Guide Formative Assessment - Work with table mate.

Tuesday, September 9

Scientific Principles

Standards

PSc.3.1.3 - North Carolina

Explain work in terms of the relationship among the applied force to an object, the resulting displacement of the object and the energy transferred to an object.

PSc.2.1.2 - North Carolina

Explain the phases of matter and the physical changes that matter undergoes.

PSc.2.1.3 - North Carolina

Compare physical and chemical properties of various types of matter.

Essential Questions

How do I tell the difference between the states of matter? [solid, liquid, gas and plasma]

Warm Up

Study quietly- If have ?, ask.

Objective

Scholars will

PSc 2.1.1 Identify the states of matter [solid, liquid, gas, plasma]

PSc 2.1.2 Understand and diagram the physical changes of matter.

Activity 1

Discuss Ch16.2 with guided notes

Activity 2

Practice 16.2

Assessment

Quest at beginning of class

Wednesday, September 10

Pressure

Standards

PSc.3.1.3 - North Carolina

Explain work in terms of the relationship among the applied force to an object, the resulting displacement of the object and the energy transferred to an object.

PSc.3.1.2 - North Carolina

Explain the Law of Conservation of Energy in a mechanical system in terms of kinetic energy, potential energy and heat.

Essential Questions

How do different forces that act upon an object.

How do I tell the difference between the states of matter? [solid, liquid, gas and plasma]

Warm Up

Formative Probe green spine p.77

Objective

Scholars will

PSc 2.1.1 Identify the states of matter [solid, liquid, gas, plasma]

Activity 1

Discuss Ch 16.3 with guided notes

Activity 2

Demos

Assessment

Formative Probe p.77

Questions during Demos and class.

Thursday, September 11

Atoms

Standards

PSc.2.2.1 - North Carolina

Infer valence electrons, oxidation number, and reactivity of an element based on its location in the Periodic Table.

PSc.2.1.4 - North Carolina

Interpret data presented in Bohr model diagrams and dot diagrams for atoms and ions of elements 1 through 18.

Essential Questions

How is the composition of the atom related to Chemistry in general?

Why are atoms important?

How do we know so much about atoms if we can not see them?

Warm Up

Study quietly/ Ask teacher questions

Objective

Scholars will

I will understand the atom both conceptually and mathematically.

I will understand and be able to draw Bohr models and Lewis dot diagrams.

Activity 1

Hand out Atom Project

Resources:

- Atom Project.odt: <https://www.filepicker.io/api/file/08JzxmUDRcGg3lxXCejT>

Activity 2

Unit Test

Assessment

Unit Test Summative Assessment

Activity 3

Begin Discussion of Atom

Friday, September 12

Atom continued

Standards

PSc.2.2.1 - North Carolina

Infer valence electrons, oxidation number, and reactivity of an element based on its location in the Periodic Table.

PSc.2.1.4 - North Carolina

Interpret data presented in Bohr model diagrams and dot diagrams for atoms and ions of elements 1 through 18.

Essential Questions

How is the composition of the atom related to Chemistry in general?

Why are atoms important?

How do we know so much about atoms if we can not see them?

Warm Up

Formative Probe "Iron Bar" green spine p.17

Objective

Scholars will

I will understand the atom both conceptually and mathematically.

I will understand and be able to draw Bohr models and Lewis dot diagrams.

Activity 1

Discuss Atom with Guided notes

Activity 2

Atoms Family Activity skit

Assessment

EXTRA

Crash Course Chem = Atom

Resources:

- Crash Course Chem = Atom: <https://www.youtube.com/watch?v=thnDxFdkzZs>

